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2013

(May)

BIOTECHNOLOGY

(General)

Course: 201

(Biophysics and Analytical Technique)

Full Marks: 48

Pass Marks: 19

Time: 2 hours

The figures in the margin indicate full marks for the questions

Choose the correct answer :

 $1 \times 4 = 4$

- (a) Atomic structure of a macromolecule in solution can be determined by
 - (i) electrophoresis
 - (ii) X-ray crystallography
 - (iii) IR-spectroscopy
 - (iv) gel filtration

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thesis, photolysis of water	(3)
(b) In photosynthesis, photolysis of water take place at (i) photosystem I (ii) photosystem II (iii) thylakoid membrane (iv) None of the above (c) The basic principle of NMR is the	2. Write on the following: (a) Gel filtration 2 (b) Lambert-Beer law 3 (c) Redox potential 3 (d) Photoreceptors 3
(c) The basic principle (i) nuclear spin giving rise to nuclear spin resonance (ii) electronic spin giving rise to electronic spin resonance (iii) electronic spin giving rise to nuclear spin resonance (iv) nuclear spin giving rise to electronic spin giving rise to electronic spin resonance	3. Describe gel electrophoresis and mention the application of this technique. 7+4 Or What is the principle of partition chromatography? Describe a method using this principle of chromatography for separation of compounds. 4+7
(d) Cation exchange resin in ion exchange chromatography bears (i) negative charge (ii) positive charge (iii) both the charge (iv) no charge	4. Discuss Calvin cycle mentioning the involvement of intermediates. How many ATP and NADPH are consumed in converting CO ₂ to the level of a hexose? Or Rhodopsin is the photoreceptor protein present in eyes. How light stimulate this photoreceptor? Explain.

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(Continued)

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7+4

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5. Define and explain the enthalpy, entropy and free energy of a system. The change in free energy (ΔG) of a reaction is an important criterion of whether the reaction can occur spontaneously. Explain.

Or

Describe the basic components of a UV-Vis spectrophotometer and mention the application of this instrument.

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